

# Kinetic Nucleation in Thermal Non-Equilibrium

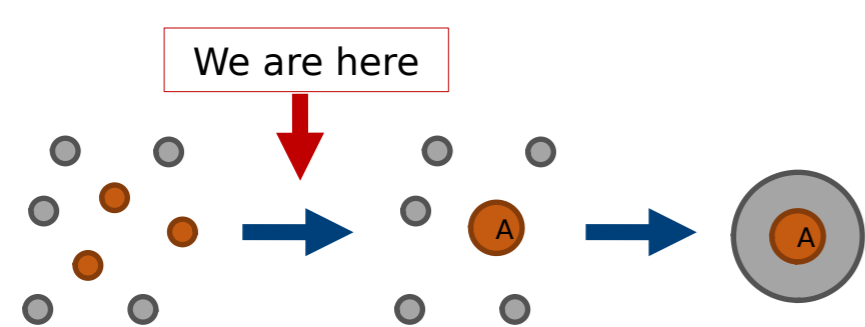
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## Take home

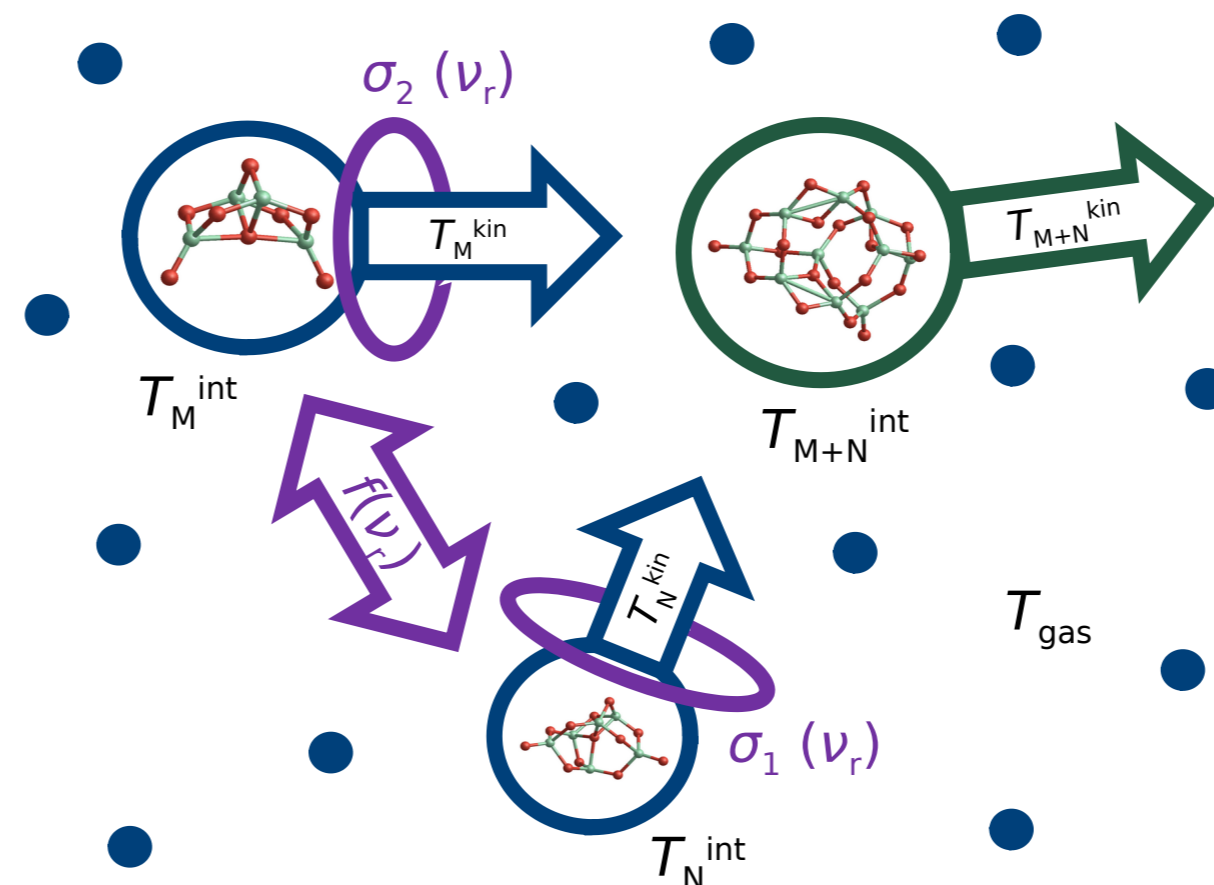
If thermal non-equilibrium is present, kinetic nucleation is affected by it. Nonetheless, the assumption of thermal equilibrium is generally justified for exoplanet atmospheres.

## Connection to Exoplanets

Clouds form when materials (●) condense onto aerosols (⊙). In gaseous exoplanets, aerosols must form from the gas phase (●) via kinetic nucleation. With this work we look at the effect of thermal non-equilibrium.



## Kinetic Nucleation

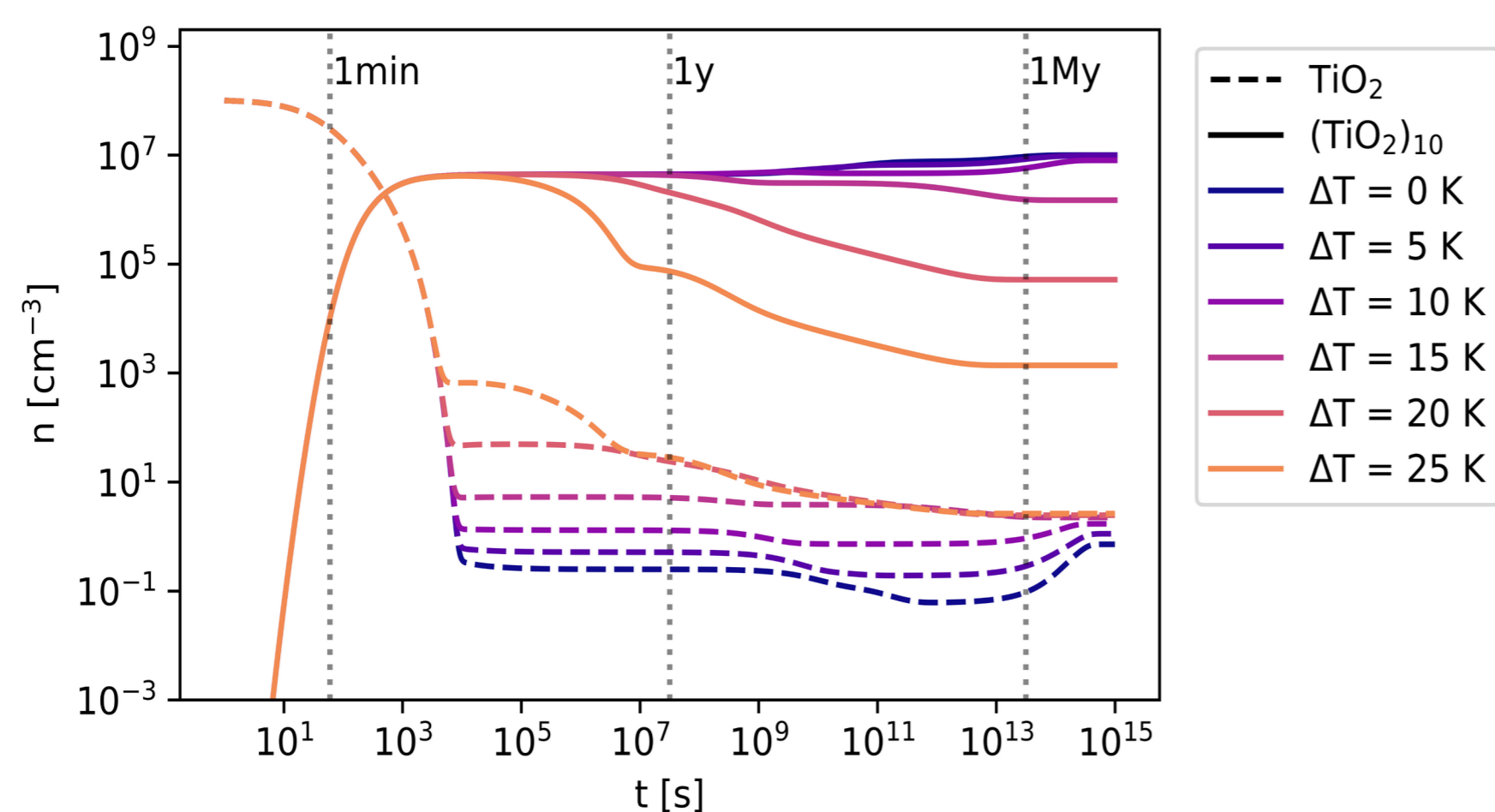


### Growth reaction rate $k^+$ :

- $T^{\text{int}}$  internal temperature
- $T^{\text{kin}}$  kinetic temperature
- $T_{\text{gas}}$  gas temperature
- $N, M, N+M$  cluster sizes
- $v_r$  relative velocity between colliding particles
- $\sigma_j(v_r)$  reaction cross section
- $f(v_r)$  velocity distribution

$$k_j^+ = \int_0^\infty \sigma_j(v_r) v_r f(v_r) dv_r$$

## Results



### Kinetic thermal non-equilibrium

$$T_{\text{gas}} = T_n^{\text{int}} \neq T_n^{\text{kin}}$$

### Internal thermal non-equilibrium

$$T_{\text{gas}} = T_n^{\text{kin}} \neq T_n^{\text{int}}$$

### Assumptions for this example:

- $\text{TiO}_2$  nucleation in a  $\text{H}_2$  gas at  $T_{\text{gas}} = 1000 \text{ K}$
- Initial number density  $n_{\text{TiO}_2} = 10^8 \text{ cm}^{-3}$
- Kinetic thermal non-equilibrium  $T_n = T_n^{\text{kin}} = T_n^{\text{int}}$
- Temperature offset  $\Delta T = T_{(\text{TiO}_2)_{10}} - T_{\text{TiO}_2}$

## Conclusions

- Thermal non-equilibrium can enhance or reduce  $(\text{TiO}_2)_{10}$  formation.
- Kinetic nucleation in hot, low-density environments (e.g. AGB stars [5]) can be affected by thermal non-equilibrium.
- Thermal equilibrium is a good assumption for exoplanet atmospheres.

## Get in Touch!



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### References

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Background: tirachard via freepick

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